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Diastereoselective 1,3-dipolar cycloaddition of pyrylium ylides with chiral enamides†

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Diastereoselective 1,3-dipolar cycloaddition of pyrylium ylides with chiral enamides†

Introduction

1,3-Dipolar cycloaddition of pyrylium ylides 1 with activated alkenes is a powerful synthetic tool providing direct access to the 8-oxa-bicyclo[3,2,1]octane motif 2 (Fig. 1).1–3 Common in a number of biologically active natural products such as englerin A,5 3 and aspergillus PZ,6 the oxygen bridged bicycle 2 has potential as a scaffold for the development of pharmaceuticals.

Fig. 1 1,3-Dipolar cycloaddition approach to 8-oxa-bicyclo[3,2,1]-octanes.

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†Electronic supplementary information (ESI) available: 1H and 13C spectra for cycloadducts 6a–i and acetamide 8, CCDC reference number 856583. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c2ob07007c

Our group became involved in finding new applications for this interesting reaction,7 with particular attention drawn to the potential as a scaffold for the development of pharmaceuticals.

Results and discussion

All starting materials for this study were prepared according to literature procedures. Acetoxy pyrones 4a–c were prepared from corresponding furyl alcohols5 or fructose.10 Enamides 5a, 5f and 5g were prepared via palladium catalysed trans-vinylations,11 ynamide 5e and enamides 5b–d and 5h were prepared by copper catalysed cross-couplings.12

At first we decided to establish the best cycloaddition conditions for reactions with non-chiral enamides (Table 1). Acetoxy pyrone 4a was reacted under triethyl amine promoted conditions (A)2 and thermal conditions (B).1

Under both sets of conditions the reactions with the simplest enamide 5a proceeded smoothly to give the desired product 6a. The yield was marginally better under thermal conditions, although this reaction required a significantly longer time to reach complete conversion of the starting material. The slightly lower yield of the base-promoted reaction was due to formation of the pyrylium ylide dimer 72 as a minor byproduct. Reaction of the trans-ynamide 5b produced the product 6b but in a lower yield, while the cis-ynamide 5c gave no detectable cycloaddition. Cycloaddition of the gem-disubstituted alkene 5d gave a different product, which decomposed upon attempts at purification by column chromatography. Our tentative assignment of the structure 6e was made on the basis of the crude 1H NMR analysis, which showed the absence of the enone double bond, and instead the presence of resonances corresponding to the double bond of the cyclic enol ether. Formation of this regiosomer is unusual and will be further investigated in a separate study.

The ynamide 5e gave no cycloaddition products under either thermal or base catalysed conditions.

Having established that the mono-substituted double bond gave the best yields of the cycloadduct, we proceeded to
investigate the reactions of chiral enamides 5f–h under base-promoted milder reaction conditions (Table 2).

Gratifyingly, cycloaddition reactions of terminal enamides 5f–h proceeded in moderate to good yields and the products were formed as single diastereoisomers. Chiral trans-enamide 5i gave cycloadduct 6i in a lower yield as was expected from our initial experiments. The relative stereochemistry of the cycloadducts was established via X-ray analysis of 6d (Fig. 2) and further correlation of the NMR data for the products.

The stereochemical outcome of the reactions can be rationalized by the model presented in Fig. 3. The enamide assumes the preferred conformation and the endo cycloaddition happens on the least hindered face of the alkene giving rise to the observed stereochemistry of the adducts.

To conclude this study we decided to demonstrate possible cleavage of the chiral auxiliary. Cycloadduct 6f was reduced on a continuous flow hydrogenation setup (H-Cube) to give both reduction of the enone double bond and cleavage of the chiral auxiliary. We discovered that direct purification of the primary amine by column chromatography was problematic due to its high polarity and the crude product was converted into an acetamide 8, which was isolated in a 50% yield after two steps (Scheme 1).

### Table 1  Cycloadditions with simple enamides 5a–e

<table>
<thead>
<tr>
<th>Enamide</th>
<th>Conditions</th>
<th>Time (d)</th>
<th>Product</th>
<th>Yield</th>
</tr>
</thead>
<tbody>
<tr>
<td>5a</td>
<td>A</td>
<td>1.5</td>
<td>6a</td>
<td>72%</td>
</tr>
<tr>
<td>5a</td>
<td>B</td>
<td>4</td>
<td>6a</td>
<td>85%</td>
</tr>
<tr>
<td>5b</td>
<td>A</td>
<td>2</td>
<td>6b</td>
<td>32%</td>
</tr>
<tr>
<td>5c</td>
<td>A</td>
<td>2</td>
<td>—</td>
<td>—</td>
</tr>
<tr>
<td>5d</td>
<td>A</td>
<td>1.5</td>
<td>6c</td>
<td>ca. 25% (decomposed on purification)</td>
</tr>
<tr>
<td>5e</td>
<td>A or B</td>
<td>2</td>
<td>—</td>
<td>—</td>
</tr>
</tbody>
</table>
Conclusions

In summary, we have demonstrated the first example of the use of amides as alkene activators in the 1,3-dipolar cycloaddition reactions of pyrylium ylides. The use of chiral oxazolidinones as chiral auxiliaries leads to highly stereoselective cycloadditions and the first asymmetric synthesis of the bicyclic amine 8.

Table 2  Cycloadditions with chiral enamides 5f–i

<table>
<thead>
<tr>
<th>Pyrone</th>
<th>Enamide</th>
<th>Temp</th>
<th>Time (h)</th>
<th>Product</th>
<th>Yield (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td><img src="image1.png" alt="Image" /></td>
<td><img src="image2.png" alt="Image" /></td>
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<tr>
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<td><img src="image5.png" alt="Image" /></td>
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<td>48</td>
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<td><img src="image8.png" alt="Image" /></td>
<td>rt</td>
<td>48</td>
<td><img src="image9.png" alt="Image" /></td>
<td>51</td>
</tr>
<tr>
<td><img src="image10.png" alt="Image" /></td>
<td><img src="image11.png" alt="Image" /></td>
<td>50 °C</td>
<td>48</td>
<td><img src="image12.png" alt="Image" /></td>
<td>63</td>
</tr>
<tr>
<td><img src="image13.png" alt="Image" /></td>
<td><img src="image14.png" alt="Image" /></td>
<td>rt</td>
<td>24</td>
<td><img src="image15.png" alt="Image" /></td>
<td>87</td>
</tr>
<tr>
<td><img src="image16.png" alt="Image" /></td>
<td><img src="image17.png" alt="Image" /></td>
<td>rt</td>
<td>48</td>
<td><img src="image18.png" alt="Image" /></td>
<td>21</td>
</tr>
</tbody>
</table>
12 h when TLC analysis indicated complete consumption of the starting material 4. Solvents and volatiles were removed under reduced pressure and the residue was purified by column chromatography (40–60 petroleum ether–ethyl acetate 3 : 1) to give the cycloadducts 6.

Thermal cycloadditions (method B)

A solution of acetoxy pyrones 4 (1.28 mmol) and enamides 5 (6.4 mmol) in toluene (7 mL) was placed in a sealed tube containing a Teflon stirrer bar. The sealed tube was flushed with N2, sealed with a Teflon screw-cap and heated in an oil bath (150 °C) under stirring until TLC indicated complete consumption of the starting material 4 (usually after a period of 4 days). The reaction mixture was allowed to cool down, volatiles were removed under reduced pressure and the residue was purified by column chromatography (40–60 petroleum ether–ethyl acetate 3 : 1) to give the cycloadducts 6.

Cycloadduct 6a a white solid: m.p. 76 °C; \( \lambda_{\text{max}}(\text{KBr}) 1740 \) (m), 1686 (s); \( m/z \) (ES\(^+\)) found 208.0974 (MH\(^+\)) \( \text{C}_{11}\text{H}_{14}\text{NO}_{3} \) requires 208.0974; \(^1\)H (500 MHz, CDCl\(_3\)) \( \delta 1.82 \) (1H, ddd, \( J 14.0, 6.0, 1.5 \) Hz), 2.04 (2H, m), 2.40 (2H, t, \( J 8.5, 8.5 \) Hz), 2.67 (1H, ddd, \( J 14.0, 9.0, 6.0 \) Hz), 3.28 (2H, m), 4.56 (1H, d, \( J 9.0 \) Hz), 4.70 (1H, dd, \( J 10.0, 6.0 \) Hz), 5.40 (1H, dd, \( J 6.0, 4.5 \) Hz), 6.15 (1H, dd, \( J 10.0, 1.5 \) Hz), 7.22 (1H, dd, \( J 10.0, 4.5 \) Hz). \(^{13}\)C (125 MHz, CDCl\(_3\)) 0.0, 0.9, 12.6, 17.5, 28.2, 44.7, 56.3, 69.2, 109.2, 132.5, 157.8, 177.5.

Cycloadduct 6b a white solid: m.p. 90–91 °C; \( \lambda_{\text{max}}(\text{KBr}) 1752 \) (s), 1697 (m); \( m/z \) (ES\(^+\)) found 221.1121 (MH\(^+\)) \( \text{C}_{12}\text{H}_{16}\text{NO}_{4} \) requires 221.1121; \(^1\)H (500 MHz, CDCl\(_3\)) \( \delta 2.10 \) (1H, d, \( J 6.0 \) Hz), 2.27 (1H, m), 2.41 (2H, t, \( J 9.0 \) Hz), 4.13 (1H, app t, \( J 6.0 \) Hz), 5.13 (1H, dd, \( J 6.0, 5.0 \) Hz), 6.11 (1H, dd, \( J 10.0, 1.5 \) Hz), 7.21 (1H, dd, \( J 10.0, 5.0 \) Hz). \(^{13}\)C (125 MHz, CDCl\(_3\)) 0.0, 0.9, 12.6, 17.5, 28.2, 44.7, 56.3, 69.2, 109.2, 132.5, 157.8, 177.5.

Cycloadduct 6c a white solid: m.p. 131 °C; \( [\alpha]_{D}^{20} = +370.1 \) (c = 0.95, CHCl\(_3\)); \( \lambda_{\text{max}}(\text{KBr}) 1750 \) (s), 1696 (s); \( m/z \) (ES\(^+\)) found 300.1236 (MH\(^+\)) \( \text{C}_{12}\text{H}_{16}\text{NO}_{4} \) requires 300.1236; \(^1\)H (500 MHz, CDCl\(_3\)) \( \delta 2.10 \) (1H, d, \( J 6.0 \) Hz), 2.27 (1H, m), 2.41 (2H, t, \( J 9.0 \) Hz), 4.13 (1H, s), 4.23 (1H, app t, \( J 6.0 \) Hz), 5.13 (1H, dd, \( J 6.0, 5.0 \) Hz), 6.11 (1H, dd, \( J 10.0, 1.5 \) Hz), 7.21 (1H, dd, \( J 10.0, 5.0 \) Hz). \(^{13}\)C (125 MHz, CDCl\(_3\)) 0.0, 0.9, 12.6, 17.5, 28.2, 44.7, 56.3, 69.2, 109.2, 132.5, 157.8, 177.5.

Cycloadduct 6d a white solid: m.p. 168–170 °C; \( [\alpha]_{D}^{20} = +8.8 \) (c = 1.0, CHCl\(_3\)); \( \lambda_{\text{max}}(\text{KBr}) 1755 \) (s), 1708 (s), 1409 (s); \( m/z \) (ES\(^+\)) found 286.1089 (MH\(^+\)) \( \text{C}_{16}\text{H}_{18}\text{NO}_{4} \) requires 286.1079; \(^1\)H (400 MHz, CDCl\(_3\)) \( \delta 1.68 \) (1H, dd, \( J 13.5, 1.5 \) Hz), 2.28 (1H, ddd, \( J 13.5, 10.0, 8.5 \) Hz), 4.02 (1H, ddd, \( J 14.0, 9.0, 8.5 \) Hz), 4.09 (1H, app s), 4.10 (1H, app s), 4.20 (1H, ddd, \( J 10.0, 6.0, 4.5 \) Hz), 4.50 (1H, ddd, \( J 8.5, 1.5 \) Hz), 5.50 (1H, dd, \( J 6.0, 4.5 \) Hz), 6.10 (1H, dd, \( J 10.0, 1.5 \) Hz), 7.10 (1H, d, \( J 10.0 \) Hz), 7.3 (5H, m). \(^{13}\)C (125 MHz, CDCl\(_3\)) 29.7, 37.9, 55.7, 58.9, 66.6; 74.9, 78.0, 127.5, 127.7, 129.1, 129.2, 135.1, 151.1, 157.7, 196.4.

Cycloadduct 6e a white solid: m.p. 195–199 °C; \( [\alpha]_{D}^{20} = +25.9 \) (c = 2.8, CHCl\(_3\)); \( \lambda_{\text{max}}(\text{KBr}) 1742 \) (s), 1685 (s), 1400 (s); \( m/z \) (ES\(^+\)) found 384.1230 (MNa\(^+\)) \( \text{C}_{22}\text{H}_{19}\text{NO}_{4}\text{Na} \) requires 384.1236.

Experimental section

All reactions were carried out under an inert atmosphere (N\(_2\)) unless stated otherwise. Reaction solvents (toluene, CH\(_2\)Cl\(_2\)) were freshly distilled from calcium hydride under a N\(_2\) atmosphere prior to use. All NMR spectral data was collected on Bruker DRX-500 or Bruker AV-400 spectrometers. IR spectra were collected on a Perkin-Elmer Spectrum One FT-IR spectrometer, optical rotations were measured on a Perkin-Elmer Model 341 polarimeter and mass-spec data was collected on a Waters GCT-Premier spectrometer with NanoMate attachment.

Base promoted cycloadditions (method A)

To a stirred solution of acetoxy pyrones 4 (1.28 mmol) and enamides 5 (6.4 mmol) in DCM (5 mL) was added triethylamine (696 µL, 5.12 mmol) in three equal portions over a period of 36 h at room temperature. Stirring was continued for another 4218 | Org. Biomol. Chem., 2012, 10, 4215–4219

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Cartridge at 80 °C on the H-Cube hydrogenation setup working in full hydrogenation mode until TLC indicated disappearance of compounds running mid-plate in hexane–EtOAc 2:1 mixture. Solvents were removed under vacuum and to the residual ammonium acetate salt was added DCM (10 mL), Et3N (0.5 mL, excess) and acetic anhydride (0.2 mL, excess). The resulting mixture was stirred for 6 h and concentrated. The residue was purified by flash column chromatography (40–60 petrol–EtOAc 3:1) to give the product 8 (17.4 mg, 50%) as a colourless oil: [α]D20 = +115.5 (c = 1.01, CHCl3); λmax (thin film) 3490 (br m), 2950 (m), 1720 (s), 1650 (s); m/z (ES+) found 184.0981 (MH+) C3H14NO3 requires 184.0974; 1H (500 MHz, CDCl3) δ 1.68 (1H, dd, J 14.0, 1.5 Hz), 2.09 (3H, s), 2.13 (1H, m), 2.28 (1H, dd, J 13.0, 7.5 Hz), 2.41 (1H, dd, J 17.0, 8.0 Hz), 2.58–2.76 (2H, m), 4.20 (1H, d, J 8.0 Hz), 4.44 (1H, app t, J 6.5 Hz), 4.52 (1H, dd, J 13.0, 10.0, 6.5 Hz). 13C (125 MHz, CDCl3) 3.35, 15.7, 20.4, 53.8, 58.8, 63.8, 157.2, 197.9.

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Notes and references

3 For a recent review and examples see: V. Singh, U. M. Krishna, V. Vikrant and G. K. Trivedi, Tetrahedron, 2008, 64, 3405.