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Biphenyl dioxygenase-catalysed cis-dihydroxylation of tricyclic azaarenes: chemoenzymatic synthesis of arene oxide metabolites and furoquinoline alkaloids†

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Biotransformation of acridine, dictamine and 4-chlorofuro[2,3-b]quinoline, using whole cells of Sphingomonas yanoikuyae B8/36, yielded five enantiopure cyclic cis-dihydrodiols, from biphenyl dioxygenase-catalysed dihydroxylation of the carbocyclic rings. cis-Dihydroxylation of the furan ring in dictamine and 4-chlorofuro[2,3-b]quinoline, followed by ring opening and reduction, yielded two exocyclic diols. The structures and absolute configurations of metabolites have been determined by spectroscopy and stereochemical correlation methods. Enantiopure arene oxide metabolites of acridine and dictamine have been synthesised, from the corresponding cis-dihydrodiols. The achiral furoquinoline alkaloids robustine, γ-fagarine, haplopine, isohaplopine-3,3′-dimethylallyl ether and pteleine have been obtained, from either cis-dihydrodiol, catechol or arene oxide metabolites of dictamine.

Further metabolism of the benzo[b]furan 2,3-cis-diols involved spontaneous ring opening and enzyme-catalysed carbonyl reduction to give exocyclic phenolic diol products (Scheme 1b).2c

Dihydroxylation of the 3,4-bond in the electron-deficient pyridine ring of the quinoline substrates was found to yield only minor metabolites in comparison with its carbocyclic 5,6- and 7,8-bonds. However, when benzo[b]thiophenes and benzo[b]furan substrates, containing electron-rich heteroaromatic rings, were used as substrates, dihydroxylation of the 2,3-bond revealed a more favourable metabolic route (Schemes 1a and 1b). The steric dimensions of the active site in TDO, expressed in P. putida UV4, limited the acceptable size of substrates to mono- or bi-cyclic arenes (Schemes 1a and 1b). However, the biphenyl dioxygenase (BPDO) enzyme, present in the B8/36 mutant strain of Sphingomonas yanoikuyae, has a larger active site and was able to accept tri-, and tetra-cyclic arenes (e.g. benzo[f]quinoline, benzo[h]quinoline, phenanthidine,2a benzo[c]phenanthidine,3b Scheme 2) as substrates. It is noteworthy that in these examples a marked regioselective preference for cis-dihydroxylation was found at a bond within the bay-region.

As part of an earlier programme2b,c to investigate the ability of BPDO to catalyse the cis-tetrahydroxylation of larger polycyclic aromatic rings, it was found that bis-cis-dihydrodiols were formed as further metabolites of the initial cis-dihydrodiols derived from larger carbocyclic (e.g. anthracene, chrysene, benzo[a]anthracene) and heterocyclic (e.g. acridine, phenazine, benzo[b]naphtha[2,1-d]thiophene) substrates. The

Introduction

Polycyclic azaarenes are ubiquitous in the environment as atmospheric pollutants, resulting from incomplete combustion of nitrogen-containing molecules present in fossil fuels or tobacco and also as plant alkaloids.1a,b Some larger members of the family of aza-polycyclic aromatic hydrocarbons (APAHs) present a significant hazard to human health, resulting from the mutagenicity/carcinogenicity of their mammalian metabolites.1c–e The mineralization of APAHs and alkaloids containing azaaaromatic rings by soil bacteria, via non-mutagenic/non-carcinogenic metabolites can, therefore, play a useful role in reducing this problem. Earlier bacterial studies from these laboratories have focused on the toluene dioxygenase (TDO)-catalysed biodegradation of bicyclic heterocycles including quinolines,2a,b benzo[b]thiophenes2c and benzo[b]furans,2c using the UV4 mutant strain of Pseudomonas putida (Schemes 1a and 1b). Regioselective cis-dihydroxylation of the carbocyclic and the heterocyclic rings in the quinolines (5,6 and/or 7,8 and/or 2,3 bonds), benzo[b]thiophenes (4,5 and/or 2,3 bonds) and benzo[b]furans (6,7 and/or 2,3 bonds), occurred to give the corresponding cis-dihydrodiol metabolites. The 3-hydroxyquinoline and anthranilic acid metabolites of quinoline were assumed to be derived from the undetected heterocyclic cis-3,4-dihydrodiol intermediate (Scheme 1a).2a,b

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similarity in size and shape of the linear tricyclic arenes, anthracene and acridine, and their acceptability as substrates for the BPDO enzyme, prompted this comparative biotransformation study of acridine with furo[2,3-b]quinoline substrates. Following our earlier reports on the isolation and synthesis of quinoline alkaloids, from plants of the Rutaceae family, e.g. Choisya ternata, and Skimmia japonica, linear furoquinolines (4-chlorofuro[2,3-b]quinoline and dictamnine) were briefly examined as potential substrates, using whole cells of S. yanoikuyae B8/36 expressing BPDO enzyme.4

In our preliminary studies of the biotransformations of acridine and dictamnine, using S. yanoikuyae B8/36, we had reported the presence of the corresponding cis-dihydrodiol metabolites. This comprehensive study now provides full structural and stereochemical characterization of all new bacterial metabolites and shows how they can be utilized in
the chemoenzymatic synthesis of a wider range of animal and plant metabolites, e.g. arene oxides and furoquinoline alkaloids.

Results and discussion

(i) Biotransformation of acridine 1

The mammalian metabolism and mutagenicity of acridine 1 have been studied over many years using dog, rabbit and rat liver cells. The major metabolites were found to be 2-hydroxyacridine, 9-acridone, 2-hydroxy-9-acridine and trans-1,2-dihydroxy-1,2-dihydroacridine 3 (Scheme 3). It is probable that trans-dihydrodiol 3 and 2-hydroxyacridine were derived from the undetected acridine 1,2-oxide 2. The identification of these mammalian metabolites of acridine 1, which could be accounted for, mainly, by monooxygenase-catalysed oxidation, prompted the preliminary and current study of its dioxygenase-catalysed metabolism.

Biotransformation of acridine 1, using S. yanoikuyae B8/36 under similar conditions to those used for other azaarene substrates, followed by extraction (EtOAc) and column chromatography, yielded cis-dihydrodiol 4 ([α]D 71) in acceptable yield (42%). The structure of cis-diol 4 was determined by NMR, MS and elemental microanalysis. The enantiomeric excess value (ee) was estimated as >98% by reaction with (R)-(+)- and (S)-(−)-2-(1-methoxyethyl)phenylboronic acid (MPBA) and 1H-NMR analysis of the resulting boronates.

The absolute configuration of cis-dihydrodiol 4 was initially assigned as (1R,2S), based on the well established 1H-NMR pattern previously observed for MPBA derivatives from other polycyclic arene cis-dihydrodiol metabolites (e.g. from naphthalene, anthracene, phenanthrene and their aza-analogues). The observation of a larger chemical shift value (δH 3.18) for the MeO group protons, using the (R)(+)-MPBA compared with the value obtained using (S)(−)-MPBA (δH 3.11), was again assumed to be consistent with a benzylic (R) and an allylic (S) configuration for cis-dihydrodiol 4. The reliability of the MPBA method for the linear azaarene cis-dihydrodiol 4 was confirmed by an unequivocal stereochemical correlation sequence similar to that used for other polycyclic arene cis-dihydrodiols (Scheme 4). The sequence involved a catalytic hydrogenation (H2/Pd/C) to yield cis-tetrahydrodiol 5 followed by bis-acetylation (Ac2O, pyridine) to give cis-diacetate 6. In the final step, an oxidative ring opening reaction ([RuO2/NaIO4] gave a mixture of dicarboxylic acid products. It was assumed that the bicyclic dicarboxylic acid 7 was formed initially and then a part of it degraded to acyclic dicarboxylic acid 8 via a further oxidative ring opening reaction. The mixture of dicarboxylic acids 7 and 8 was methylated (CH2N2) to yield dimethylesters 9 and 10 which were separated by column chromatography. The minor component, dimethyl(2,3-diacetoxy)adipate 10 ([α]D −14) was of established (2S,3S) configuration and thus the (1R,2S) configuration was unequivocally assigned to (+)-cis-dihydrodiol 4.

It has been proposed that the mutagenicity/carcinogenicity associated with some larger PAHs and APAHs results from: (i) a monoxygenase-catalysed epoxidation of a carbocyclic ring to yield an arene oxide (cf. compound 2), (ii) an epoxide hydrolyse-catalysed hydrolysis of the arene oxide to yield a trans-dihydrodiol (cf. compound 3), (iii) a monoxygenase-catalysed epoxidation of the alkene bond in the trans-dihydrodiol to yield diastereoisomeric trans-diol epoxides and (iv) nucleophilic attack of DNA on the epoxide ring within a bay region to yield a covalent adduct. Although the corresponding acridine trans-diol epoxides from metabolite 3 could, in principle, also be mutagens, their synthesis and mutagenicity has not yet been reported.

(ii) Biotransformation of furoquinolines 11–13

In common with acridine 1, the mammalian metabolism and mutagenicity of dictamine 12 and other furoquinoline alkaloids, e.g. γ-fagarine, had been reported earlier. In a more recent study, from these laboratories, the furoquinoline...
alkaloid skimmianine 13 was found to be the major compound present in *C. ternata* and was thus available as a potential substrate for the current biotransformation studies. However, dictamnine 12, another furoquinoline alkaloid required as a potential substrate, was not isolated from *C. ternata*. Thus, a five-step chemical synthesis of dictamnine 12 was carried out, starting from aniline and using the literature procedure which involved 4-chlorofuroquinoline 11 as precursor. Furoquinolines 11–13 were thus also available as possible substrates for BPDO.

Furoquinolines 11–13 were added, individually, as substrates to *S. yanoikuyae* B8/36, under the conditions used previously for the successful biotransformation of acridine 1. The results, shown in Scheme 5, indicate that while 4-chlorofuro[2,3-b]quinoline 11 and dictamnine 12 each yielded a mixture of two *cis*-dihydrodiol products resulting from BPDO-catalysed *cis*-dihydroxylation of the 5,6 and 7,8 bonds of the carbocyclic ring, skimmianine 13 was not an acceptable substrate. The mixtures of metabolites (14/15 and 16/17) were separated into individual *cis*-dihydrodiols by PLC purification. The structures of diol metabolites (14–17) were established by analyses of NMR and MS data while the ee values (>98%) and absolute configurations were again determined by formation of the corresponding diastereomeric MPBA esters and their analysis by 1H-NMR spectroscopy. As found for the acridine *cis*-dihydrodiol 4, the larger chemical shift values ($\delta_H$) for the exocyclic MeO group of *cis*-dihydrodiols 14–17, ($\delta_H$ 3.23–3.25) using the (R)-(+) MPBA compared with ((S)-(2)-MPBA 14–17, ($\delta_H$ 3.19–3.20) were consistent with benzylic (R) and allylic (S) configurations in each case.

While *cis*-dihydroxylation had occurred exclusively at the 1,2 bond of acridine 1, similar regioselectivity for the equivalent

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**Scheme 4**

![Scheme 4](image)

**Scheme 5**

![Scheme 5](image)
(5,6) bond in furoquinolines 11 and 12 was not found. A modest preference (38% yield) was observed for BPDO-catalysed cis-dihydroxylation at the 7,8-bond to form cis-diol 15 compared with the 5,6-bond (10% yield) to give cis-diol 14, when 4-chlorofuro[2,3-b]quinoline 11 was the substrate. A stronger preference for oxidation of the 7,8-bond was found with dictamnine 12 as the substrate, which resulted in cis-diol 17 being the major metabolite (20–30% yield) relative to cis-diol 16 (1–3% yield). The combined isolated yields (21–33%) of dictamnine cis-dihydrodiols (16 and 17) were slightly lower than 4-chlorofuroquinoline cis-dihydrodiols (14 and 15, 48%); no cis-dihydrodiol metabolites were detected from skinni-nine 13 as substrate. These observations suggest that the presence of substituents at C-4, C-7 and C-8 and the overall steric requirements of the substrate within the active site of the BPDO enzyme are important factors. Based on isolated yields, it appears that cis-dihydroxylation occurred preferentially at the less sterically hindered 7,8-bond and that the best yields resulted from the use of the smaller substrates (11 and 12). As the largest substrate, skinni-nine 13, did not yield cis-diol metabolites, this is consistent with its failure to be accommodated within the BPDO active site. However, alternative factors, including aqueous solubility, toxicity and further metabolism, could influence the isolated yields of bioproducts.

The most polar metabolites, formed from 4-chloro-furo[2,3-b]quinoline 11 and dictamnine 12, were found to be exocyclic diols (compounds 20 and 23) but were isolated in very low yields (1–2%, Scheme 5). While the structures of optically active diols 20 and 23 were assigned by NMR and MS spectroscopic analysis, their ee values and absolute configurations were not determined. It was assumed, that the exocyclic diols 20 and 23, resulted from: (a) BPDO-catalysed cis-dihydroxylation at the 2,3-bond to give transient intermediates 18 and 21, (b) reversible ring opening of these hemiacetals (cf. mutarotation) to yield the undetected aldehydes 19 and 22, (c) epimerization, following reversible ring closure, to yield a mixture of the corresponding cis- and trans-dihydrodiols, and (d) carbonyl reductase-catalysed (CRED) reduction of the aldehyde group in intermediates 19 and 22. A similar sequence of TDO-catalysed cis-dihydroxylation of the furan ring of benzo[b]furans, spontaneous equilibration via a reversible ring opening process to yield the corresponding phenolic aldehydes and CRED-catalysed reduction of the resulting aldehyde group was earlier assumed to account for the isolation of the exocyclic diols shown in Scheme 1 (b).24

The origin of mutagenicity associated with dictamnine 13 has not yet been rigorously established.25–27 However, it has been proposed that, in common with other naturally occurring mutagenic furans, e.g. aflatoxin B1 and 8-methoxypsoralen, the corresponding transient furan epoxides,27 formed as initial mammalian metabolites via monooxygenase-catalysed epoxidation, e.g. arene oxide 24 (Scheme 5) may be responsible for their mutagenicity. It has been proposed that the mutagenicity results from the ability of furan epoxides to form covalently bound adducts following nucleophilic ring-opening reactions with DNA.27–29

(iii) Application of acridine cis-dihydrodiol 4 in the synthesis of arene oxide 2

As part of an earlier study of the mammalian metabolism and mutagenicity/carcinogenicity of PAHs and APAHs, (1R,2S)-arene oxide 2 was obtained via an eight stage chemical synthesis, involving a chemical resolution of MTPA esters, with an overall yield of ca. 13%.9 Alkaline hydrolysis (KOH, t-BuOH) of (1R,2S)-arene oxide 2 gave the mammalian metabolite (1R,2R)-trans-1,2-dihydroacridine-1,2-diol 3.9 In the current study, the possibility of a much shorter synthesis of acridine 1,2-oxide 2 was examined (Scheme 6), using the readily available bacterial metabolite, (1R,2S)-cis-1,2-dihydroacridine-1,2-diol 4. Treatment of diol 4 with 1-bromocarbonyl-1-methylethyl acetate, in acetonitrile solution, gave a mixture of bromoacetates 25/26 whose structures were determined from 1H-NMR and MS data. Due to their instability, during attempted separation, the mixture of bromoacetates 25 and 26 in Et2O solution was reacted directly with NaOMe. Using this two step method, the relatively stable (1R,2S)-arene oxide 2 was synthesised from cis-dihydrodiol 4 in 66% yield. Despite the stability of arene oxide 2, it was not detected during mammalian metabolism, probably due to its further metabolism via a rapid epoxide hydrolase-catalysed conversion to the corresponding trans-dihydrodiol 3.30 A preliminary study31 later showed that when the stable acridine cis-dihydrodiol 4 was used as a substrate for S. yanoikuyae B8/36, it was also further metabolised and formed a bis-cis-dihydrodiol bioproduct.

(iv) Application of dictamnine cis-dihydrodiols 16 and 17 as precursors in the synthesis of furoquinoline alkaloids

The potential of dictamnine cis-dihydrodiol metabolites 16 and 17 in the biomimetic synthesis of furoquinoline alkaloids, including the proposed arene oxide intermediate 27, was of biosynthetic interest (Schemes 7 and 8). Possible biosynthetic pathways to furoquinoline alkaloids occurring in Rutaceaeous plants, e.g. Skimmia japonica and Choisya ternata, have been studied using 13C-labelled precursors.106 These labelling studies showed that enzyme-catalysed hydroxylation could

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occur on the benzene ring of dictamnine \( \text{12} \) to yield a wider range of furoquinoline alkaloids e.g. skimmianine \( \text{13} \) and possibly also robustine \( \text{30} \) and \( \gamma \)-fagarine \( \text{31} \) (Scheme 7). It was proposed that skimmianine \( \text{13} \) could be formed via a monoxygenase-catalysed epoxidation of dictamnine \( \text{12} \), to yield the transient arene oxide \( \text{27} \), followed by epoxide hydrolase-catalysed hydrolysis to yield trans-dihydrodiol \( \text{28} \).10a The possibility of an alternative dioxygenase-catalysed cis-dihydroxylation of dictamnine \( \text{12} \) to yield cis-dihydrodiol \( \text{17} \) was also discussed.10a The enzyme-catalysed oxidations of trans- and cis-dihydrodiols, to yield catechols followed by O-methylation, are well established metabolic steps1 and, when allied to the earlier labelling studies,10a either type of enzymatic oxidation could account for the formation of catechol \( \text{29} \) and skimmianine \( \text{13} \). To date, none of the potential biosynthetic intermediates \( \text{17, 27–29} \) have been detected by the labelling studies using \( \text{Choisya ternata} \)10a or found among the furoquinoline alkaloids recently isolated from this4a or other plants in the Rutaceae family.10b

As expected, the B8/36 mutant strain of \( \text{S. yanoikuyae} \) did not yield catechol metabolites e.g. compound \( \text{29} \) from dictamnine \( \text{12} \) (Scheme 7). The biphenyl cis-diol dehydrogenase (DD) activity required to catalyse the dehydrogenation of cis-dihydrodiols to yield catechols, was blocked in the B8/36 strain. However, when the wild type strain of \( \text{S. yanoikuyae} \) (B1), expressing both BPDO and DD enzymes, was used with dictamnine \( \text{12} \), the only metabolite identified and isolated was cis-dihydrodiol \( \text{16} \), albeit in low yield (8%). This observation is consistent with both cis-dihydrodiols \( \text{16 and 17} \) being formed but the major metabolite \( \text{17} \) being further metabolized preferentially.

The \( \text{E. coli} \) narB recombinant strain, expressing naphthalene DD, has been used successfully to produce catechols in good yields from the corresponding monocyclic arene cis-dihydrodiols.11 Using \( \text{E. coli} \) narB and cis-dihydrodiol \( \text{15} \) as substrate, catechol \( \text{35} \) was detected by \( ^1 \text{H-NMR} \) spectroscopy but in low yield. Surprisingly, the required catechol metabolite \( \text{29} \), derived from dictamnine cis-dihydrodiol \( \text{17} \), could not be obtained using this method. However, it was possible to obtain catechol \( \text{29} \) in good yield (85%) using boron tribromide for the selective O-demethylation of skimmianine \( \text{13} \), isolated earlier4a from \( \text{Choisya ternata} \) (Scheme 8).

Convincing evidence of monoxygenase-catalysed epoxidation of dimethylallyl groups, and hydrolysis of the resulting epoxides to yield vicinal diols, is available from biosynthetic studies of quinoline alkaloids from plants of the Rutaceae family.10a,b Furthermore, monoxygenase-catalysed epoxidation of azaaranes, to yield the corresponding arene oxides, e.g. quinoline 5,6-oxide from quinoline, using liver microsomes with inhibition of epoxide hydrolase activity,12a,b provides a precedent for the formation of the elusive dictamnine arene oxide \( \text{27} \) and trans-dihydrodiol \( \text{28} \) metabolites. While the dioxygenase-catalysed cis-dihydroxylation of polycyclic azaaranes in bacteria, e.g. \( \text{S. yanoikuyae} \) B8/36, is well established (Schemes 1, 2 and 4), there appears to be little evidence of this pathway occurring in plants. Consequently, the monoxygenase-catalysed epoxidation sequence, shown in Scheme 7, is currently favoured over the dioxygenase pathway. Epoxidation,
as an initial step, can readily account for the formation of both monophenols (e.g. robustine 30), catechols (e.g. 7,8-dihydroxydictamnine 29), and their methylated derivatives (e.g. γ-fagarine 31 and skimmianine 13).

Our attempt to synthesise the proposed dictamnine arene oxide metabolite 27 from cis-dihydrodiol 17, via a two-step process similar to that used earlier for acridine arene oxide 2 (Scheme 6), was unsuccessful. This was due to compound 17 being less stable under the reaction conditions and more readily dehydrated under acid conditions to yield phenols (e.g. robustine 30). An alternative approach (Scheme 8) was adopted involving the catalytic hydrogenation (H2, Pd–C) of compound 17 to yield the stable cis-tetrahydrodiol 32 (76% yield). Treatment of diol 32 with 1-bromocarbonyl-1-methylethyl acetate gave trans-bromoacetate 33 in good yield (90%). Benzylic bromination of bromoacetate 33 [NBS, CCl4] gave an inseparable mixture of diastereoisomers 34 which was immediately treated with sodium methoxide in THF, to yield the proposed dictamnine arene oxide metabolite 27 (60% yield from compound 33). Initial attempts to purify this elusive arene oxide by PLC resulted in its aromatization to give the furoquinoline alkaloid robustine 30. Purification of (7S,8R)-dictamnine oxide 27 was achieved by careful crystallization. A sample of oxide 27 was found to survive in CDCl3 solution without decomposition, at ambient temperature over a 24 h period.

In the final phase of this study, cis-dihydrodiols 16 and 17, arene oxide 27 and catechol 29, as confirmed or proposed metabolites of dictamnine 12, were utilized as synthetic precursors of other furoquinoline alkaloids, using biomimetic methods (Scheme 8). While robustine 30 was obtained by isomerisation of arene oxide 27 under acidic conditions, the acid-catalysed dehydration of cis-dihydrodiol 17 was the preferred route. Methylation of robustine 30 with diazomethane yielded the alkaloid γ-fagarine 31. Under similar conditions, methylation of catechol 29 occurred mainly at C-8, to yield the alkaloid haplopine 36. Treatment of catechol 29 in acetone with 1-chloro-3-methylbut-2-ene in presence of K2CO3 resulted in the preferential prenylation at C-8 to yield phenol 37, which on methylation yielded the alkaloid isohaplopine-3,3'-dimethylallylether 38. Acid-catalysed dehydration of cis-dihydrodiol 16, to form phenol 39, followed by methylation, yielded the furoquinoline alkaloid pteleine 40.

**Conclusion**

The bacterial cis-dihydroxylation of acridine 1 and furoquinolines 11 and 12, catalysed by BPDO, yielded five carbo cyclic cis-dihydrodiols, (4, 14–17) and two exocyclic diols (20 and 23), derived from the transient heterocyclic diols 18 and 21. The structures and absolute configurations of most of the isolated metabolites were established by spectroscopic analysis and
stereochemical correlation methods. cis-Dihydriodiol 4 and 17
were used in the synthesis of the corresponding arene oxides,
2 and 27, which had been proposed as intermediates in
mammalian metabolism of acridine 1 and dictamine 12. cis-
Dihydriodiol 16 and 17 and arene oxide 27, as derivatives of
dictamine 12, have been used in the synthesis of a wide range
of furoquinoline alkaloids including robustine 30, γ-fagarine
31, haploine 36, isohaploine-3,3′-dimethylallylether 38 and
pteleine 40.

Experimental
1H and 13C NMR spectra were recorded on Bruker Avance 400,
DPX-300 and DRX-500 instruments. Chemical shifts (δ) are
reported in ppm relative to SiMe4 and coupling constants (J)
are given in Hz. Mass spectra were run at 70 eV, on a VG
Autospec Mass Spectrometer, using a heated inlet system.

Synthesis of dictamine 12
To a solution of 4-chlorofuro[2,3-b]quinoline 11 (3 g, 14.7
mmol) in dry methanol (120 cm3) was added sodium
methoxide (4.05 g, 75 mmol) and the mixture refluxed (2 h)
der under nitrogen. After removal of methanol from the reaction
mixture, under reduced pressure, the crude product purified by flash
chromatography (1 : 1, EtOAc : hexane) to afford dictamine
12 as a yellow solid (1.65 g, 56%); δH 4.42 (1 H, dd, J1,2 = 4.7, J2,3 = 4.8, 2-H), 4.86 (1 H, d,
J1,2 = 4.7, 1-H), 6.54 (1 H, dd, J1,2 = 4.8, J2,3 = 10.1, 3-H), 6.77 (1 H, d, J3,4 = 10.1, 4-H), 7.53 (1 H, dd,
J2,3 = 8.1, J3,4 = 6.9, 7-H), 7.68 (1 H, dd, J6,7 = 6.9, J5,6 = 8.5, 6-H), 7.90 (1 H, d, J6,7 = 8.1, 8-H), 7.95 (1 H, d, J5,6 = 8.5, 5-H), 8.28 (1 H, s, 9-H); δC (125 MHz, CDCl3) 66.6,
70.2, 126.3, 127.7, 127.9, 129.0, 129.1, 130.8, 132.1, 133.2,
136.3, 147.1, 153.0; m/z (EI): 215 (M⁺, 56%), 195 (23), 184 (100); ν3369 cm⁻¹ (OH); ECD: ɛ/nm 311 (Δδ 1.036), 257 (Δδ 1.644), 243 (Δδ 6.421), 215 (Δδ 4.647), 199.40 (Δδ 0.573).

(1R,2S)-1,2,3,4-Tetrahydroacridine-1,2-diol 5.
cis-Dihydriodiol metabolite 4 (0.5 g, 2.35 mmol) was catalytically
hydrogenated (H2, 10% Pd/C, 50 mg) in MeOH solution
(20 cm³) at atmospheric pressure and room temperature (4 h).
The catalyst was filtered off and the filtrate evaporated under
reduced pressure to yield (1R,2S)-1,2,3,4-tetrahydroacridine-
1,2-diol 5 as a semi-solid (0.420 g, 83%); [α]D +61 (c 1.00,
MeOH); (Found: M⁺ 215.0937. C13H13NO2 requires 215.0946);
δH (500 MHz, CDCl3) 2.12 (1 H, m, 3a-H), 2.32 (1 H, m, 3b-H),
3.11 (1 H, m, 4a-H), 3.36 (1 H, m, 4b-H), 4.28 (1 H, m, J3,4 = 3.5,
2-H), 4.93 (1 H, dd, J1,2 = 3.5, 1-H), 7.90 (1 H, dd, J6,7 = 8.0, J5,6 = 7.1,
7-H), 7.68 (1 H, dd, J6,7 = 7.1, J5,6 = 8.4, 6-H), 7.79 (1 H, d, J6,7 = 8.0,
8-H), 8.00 (1 H, d, J5,6 = 8.4, 5-H), 8.30 (1 H, s, 9-H); δC (125 MHz,
CDCl3) 26.3, 29.5, 68.9, 70.2, 126.0, 127.2, 127.6, 128.4, 129.8,
130.8, 132.1, 133.2, 136.3, 147.1, 153.0; m/z (EI): 215 (M⁺, 71%), 196 (18), 186
(74), 168 (75), 143 (100); ν3435 cm⁻¹ (OH).

(1R,2S)-1,2-Diacetoxy-1,2,3,4-tetrahydroacridine 6.
cis-Tetrahydroacridine 5 (0.18 g, 0.84 mmol) was acetylated with
Ac2O (1.5 cm³) in dry pyridine (0.7 cm³) solution by stirring the
mixture overnight at room temperature. Excess of pyridine was
removed under high vacuum, the residue treated with water
(10 cm³), and the aqueous mixture extracted with EtOAc (3 × 15
cm³). The combined ether extract was dried (Na2SO4) and
centrifuged to yield (1R,2S)-1,2-diacetoxy-1,2,3,4-tetrahydro-
acridine 6 as a white crystalline solid (0.22 g, 88%); m.p. 84–85
°C (EtOAc/hexane); [α]D −57 (c 0.85, MeOH); (Found: M⁺
299.1150. C13H15NO4 requires 299.1158); δH (300 MHz,
C6D5Cl3) 2.05 (3 H, s, OCOME), 2.15 (3 H, s, OCOME), 2.20 (1 H,
J3,4 = 3.1, 3a-H), 2.44 (1 H, m, 3b-H), 2.33 (1 H, m, 4a-H),
3.40 (1 H, m, 4b-H), 5.43 (1 H, m, J3,4 = 3.3, J3,3a = 3.1, 2-H), 6.32 (1 H,
J3,4 = 3.3, 1-H), 7.51 (1 H, dd, J7,8 = 8.1, J6,7 = 7.0, 7-H), 7.72 (1 H,
dd, J6,7 = 7.0, J5,6 = 8.5, 6-H), 7.79 (1 H, d, J6,7 = 8.1, 8-H), 8.02 (1 H, d,
J5,6 = 8.5, 5-H), 8.12 (1 H, s, 9-H); δC (125 MHz, CDCl3) 19.6, 19.7,
22.1, 28.8, 67.8, 68.3, 124.9, 125.3, 124.6, 126.7, 127.0, 128.9,
135.9, 146.5, 155.6, 160.9, 169.2; m/z (EI): 299 (M⁺, 69%), 197
(100); ν1741 cm⁻¹ (C=O), 1421.

Oxidative degradation of diacetate 6. Ruthenium(ii) oxide
hydrate (2 mg) was added to a biphasic mixture of the
diacetate 6 (0.1 g, 0.33 mmol) and NaNO3 (3.4 g, 16 mmol) in
a mixture of CCl4 (2 cm³), MeCN (2 cm³) and water (3 cm³). After
stirring the reaction mixture for 4 days at ambient tempera-
ture, dilute HCl (20 cm³, 1.5 M) was added and the reaction mixture extracted with EtOAc (3 × 30 cm³). The combined extract was washed with water, dried (Na₂SO₄), and concentrated under reduced pressure to yield a crude mixture of two dicarboxylic acids. The mixture was dissolved in MeOH (5 cm³) and treated (0 °C, 0.5 h) with excess of freshly prepared ethereal diazomethane solution. The solvents were evaporated, in a fume hood under a stream of nitrogen, and the crude methyl esters separated by column chromatography (25% → 75% EtO/Hexane) to yield dimethyl [5R,6S]-5,6-diacetoxy-5,6,7,8-tetrahydro-2,3-quinolinecarboxylate 9, the more polar major compound (21 mg, 17%), and [2S,3S]-dimethyl-(2,3-diacetoxy)-adipate 10 (9 mg, 9.3%) as the less polar minor compound.

Dimethyl [5R,6S]-5,6-diacetoxy-5,6,7,8-tetrahydro-2,3-quinolinecarboxylate 9. White crystalline solid; m.p. 115–117 °C (from CHCl₃); [α]D -53 (c 0.66, MeOH); (Found: C, 55.7; H, 5.1; N, 3.9. C₁₂H₁₁NO₄ requires C, 55.9; H, 5.2; N, 3.8%); (Found: M+ 365.1105. C₁₂H₁₂NO₄ requires 365.1111); δH (500 MHz, CDCl₃) 2.04 (3 H, s, OCOMe), 2.15 (3 H, s, OCOMe), 2.16 (1 H, m, J6a,6b 3.1, 7a-H), 2.35 (1 H, m, 7b-H), 3.11 (1 H, m, 8a-H), 3.25 (1 H, m, 8-H), 3.92 (3 H, s, COMe), 3.99 (3 H, s, COMe), 5.42 (1 H, m, J6a,6b 3.4, J6a,7a 3.1, 6-H), 6.13 (1 H, d, J6a,6b 3.4, 5-H), 8.11 (1 H, s, 4-H); δC (125 MHz, CDCl₃) 23.1 (12, 13, 14), 32.8 (12, 13, 14), 50.5 (12, 13, 14), 58.8, 64.5, 71.1, 71.2, 105.3, 105.4, 111.7, 112.8, 124.2, 142.2, 151.7, 156.7, 163.1; m/z (EI): 233 (M+%, 61%), 215 (9), 204 (100); ECD: ∆h nm 275.60 (∆ = -0.425), 246.10 (∆ = -2.722), 205.90 (∆ = -0.272).

3-[1,2-Dihydroxyethyl]-4-methoxy-1,2-dihydro-2-quinoline 11. Brown solid (1.1 mg, 1.6%); m.p. 137–140 °C (decomp.); Rf 0.20 (7% MeOH/CHCl₃); [α]D +8.8 (c 0.68, MeOH); (Found: M+ 235.0840. C₁₂H₁₃NO₄ requires 235.0845); δH (500 MHz, CDCl₃) 3.82 (1 H, dd, J9a,J9b 11.1, 1.1, 1.1, J1,J2 7.4, 2-H), 4.04 (3 H, s, OMe), 5.15 (1 H, dd, J9a,J9b 4.4, J9a,J9b 1.1, 1.1, 1-H), 7.30 (1 H, dd, J6a,J7a = 7.9, 6.9), 7.35 (1 H, d, J6a,J6b 8.2, 8-H), 7.57 (1H, dd, J9b,J9a 8.2, J9a,J9b 7.9, 7-H), 7.81 (1 H, d, J6a,J6b 7.9, 5-H), 11.68 (1 H, br s, NH); δC (75 MHz, CDCl₃) 62.9, 66.0, 69.0, 116.3, 118.9, 123.4, 123.5, 131.5, 137.7, 163.2, 166.2; m/z (EI): 235 (M+%, 4%), 204 (20), 122 (83), 105 (100), 77 (80), 51 (54), 43 (71).

(c) Biotransformation of chlorofuro[2,3-b]quinoline 11

A small-scale biotransformation on 4-chlorofuro[2,3-b]quinoline 11 (0.8 g, 3.93 mmol), using S. yanoikuyae B8/36, followed by concentration of the aqueous portion under reduced pressure and extraction of the concentrate by ethyl acetate yielded a mixture of products. PLC purification of the mixture (6% MeOH/CHCl₃) gave [7S,8R]-4-chloro-7,8-dihydrofuro[2,3-b]quinoline-7,8-diol 15 (0.35 g, 38%); Rf 0.6 (6% MeOH/CHCl₃), [5R,6S]-4-chloro-5,6-dihydrofuro[2,3-b]quinoline-5,6-diol 14 (93 mg, 10%); Rf 0.5 (6% MeOH/CHCl₃) and 4-chloro-[1,2-dihydroxyethyl]-1,2-dihydro-2-quinoline 20 (23 mg, 2.5%); Rf 0.4 (6% MeOH/CHCl₃).

(7S,8R)-4-Chloro-7,8-dihydrofuro[2,3-b]quinoline-7,8-diol 15. Brown crystalline solid (0.35 g, 38%); m.p. 121–123 °C (decomp., CHCl₃/Hexane); [α]D +138.1 (c 0.53, CHCl₃); (Found: M+ 237.0201. C₁₁H₁₁NO₄ requires 237.0193); δH (500 MHz, CDCl₃) 4.49 (1 H, dd, J7a,J7b 5.2, 7.2, 7-H), 4.81 (1 H, d, J7a,J7b 4.8, 8-H), 6.34 (1 H, dd, J6a,J6b 5.2, J6a,J7a 9.9, 6-H), 6.85 (1 H, d, J6a,J6b 2.4, 3-H), 7.05 (1 H, d, J6a,J7a 9.9, 5-H), 7.66 (1 H, d, J6a,J6b 2.4, 2-H); δC (125 MHz, CDCl₃) 63.8, 70.3, 104.2, 117.6, 120.3, 123.5, 127.5, 133.6, 144.0, 150.9, 159.1; m/z (EI): 239 [M+, 37%], 237 [M+, 43], 219 [M+H₂O, 15], 208 [M'+CHO, 75], 199 (100), 190, 184 (55), 156 (49), 149 (29), 128 (28); ECD: ∆h nm 316 (∆ = -0.31), 316 (∆ = -0.27), 253 (∆ = 4.12), 235 (∆ = 3.34), 207 (∆ = -2.07).

(5R,6S)-4-Chloro-5,6-dihydrofuro[2,3-b]quinoline-5,6-diol 14. Brown crystalline solid [93 mg, 10%]; m.p. 172–174 °C (decomp.); [α]D +189 (c 0.41, MeOH); (Found: M+ 237.0199. C₁₁H₁₃ClNO₃ requires 237.0193); δH (500 MHz, CDCl₃) 4.72 (1 H, br s, 6-H), 5.21 (1 H, dd, J6a,J6b 5.1, J6a,J6b 1.7, 5-H), 6.23 (1 H, d, J6a,J6b 5.2, J6a,J7a 1.7, 7-H), 6.64 (1 H, dd, J6a,J6b 10.8, J6a,J6b 2.7, 8-H), 6.86 (1 H, d, J6a,J6b 2.5, 3-H), 7.72 (1 H, d, J6a,J6b 2.5, 2-H); δC (125 MHz, CDCl₃) 66.1, 69.7, 105.3, 118.1, 123.5, 128.7, 136.8, 137.68, 145.8, 148.2, 161.7; m/z (EI): 237 [M+%, 33%], 221
4-Chloro-3-(1,2-dihydroxyethyl)-1,2-dihydro-2-quinoline 20. Brown solid (23 mg, 2%); m.p. 184–186 °C (decomp.); \([\delta]_D -5.1 \) (c 0.43, MeOH); (Found: M+CH2OH 208.0157. C10H7ClNO2 requires 208.0165); \(\delta_H (500 MHz, CDCl3) 3.76 \) (1 H, dd, J=5.6, J=5.6, 6-H), 3.87 (1 H, dd, J=5.6, J=5.6, 7.2, 2'-H), 4.51 (2 H, br, OH), 5.20 (1 H, dd, J=3.6, J=3.6, 7.1, 2'-H), 7.32 (2 H, m, 6-H, 8-H), 7.57 (1 H, m, 7-H), 8.00 (1 H, dd, J=5.6, J=5.6, 8.2, J=7.1, 1.1, 5-H); \(\delta_C (125 MHz, CDCl3) 65.8, 74.2, 117.1, 120.2, 125.0, 127.0, 129.6, 133.2, 138.7, 144.8, 164.0; m/z (EI): 210 \) (M+), 115 \(\text{ClCH2OH}, 31\% \), 208 (100), 162 (15), 89 (13), 32 (16).

7,8-diol 32. A solution of dictamnine 7,8-diol 17 (0.15 mg, 0.46 mmol) in MeOH (10 cm3) containing 3% Pd/C (15 mg) was stirred (1 h) under hydrogen atmosphere at atmospheric pressure. The reaction mixture was filtered and the solvent removed under reduced pressure to yield the crude product as a brown oil. Purification of the crude product by PLC (4\% MeOH/CHCl3), afforded the tetrahydro cis-diol 32 as a white solid (0.115 g, 76%); m.p. 149–151 °C (from EtOAc); \(\delta_H (500 MHz, MeOH/CHCl3) \) \([\delta]_D -25 \) (c 0.39, CHCl3); (Found: M+ 235.0843. C14H13NO4 requires 235.0844); \(\delta_C (500 MHz, CDCl3) 1.82–1.89 \) (1 H, m, 6'-H), 2.24–2.28 (1 H, m 6-H), 2.64 (1 H, br, OH), 2.76 (1 H, m, 5'-H), 2.86 (1 H, m, 5-H), 4.20 (3 H, s, OMe), 4.35 (1 H, m, 7-H), 4.70 (1 H, d, J=6.5, 3.3, 8-H), 6.95 (1 H, d, J=3.6, 2.6, 3-H), 7.53 (1 H, d, J=3.6, 2.6, 5-H); \(\delta_C (125 MHz, CDCl3) 18.3, 25.0, 58.6, 66.7, 74.0, 104.9, 106.5, 115.2, 142.2, 150.1, 158.5, 162.8; m/z (EI): 235 \) (M+, 30%), 206 (100), 188 (92), 163 (94), 133 (35).

7,8-diol 32. A solution of dictamnine 7,8-diol 17 (0.15 mg, 0.46 mmol) in MeOH (10 cm3) containing 3% Pd/C (15 mg) was stirred (1 h) under hydrogen atmosphere at atmospheric pressure. The reaction mixture was filtered and the solvent removed under reduced pressure to yield the crude product as a brown oil. Purification of the crude product by PLC (4\% MeOH/CHCl3), afforded the tetrahydro cis-diol 32 as a white solid (0.115 g, 76%); m.p. 149–151 °C (from EtOAc); \(\delta_H (500 MHz, MeOH/CHCl3) \) \([\delta]_D -25 \) (c 0.39, CHCl3); (Found: M+ 235.0843. C14H13NO4 requires 235.0844); \(\delta_C (500 MHz, CDCl3) 1.82–1.89 \) (1 H, m, 6'-H), 2.24–2.28 (1 H, m 6-H), 2.64 (1 H, br, OH), 2.76 (1 H, m, 5'-H), 2.86 (1 H, m, 5-H), 4.20 (3 H, s, OMe), 4.35 (1 H, m, 7-H), 4.70 (1 H, d, J=6.5, 3.3, 8-H), 6.95 (1 H, d, J=3.6, 2.6, 3-H), 7.53 (1 H, d, J=3.6, 2.6, 5-H); \(\delta_C (125 MHz, CDCl3) 18.3, 25.0, 58.6, 66.7, 74.0, 104.9, 106.5, 115.2, 142.2, 150.1, 158.5, 162.8; m/z (EI): 235 \) (M+, 30%), 206 (100), 188 (92), 163 (94), 133 (35).

(72), 219 (100), 191 (18), 156 (80); ECD: \(\lambda/\lambda_{nm} 316 \) (Δ3.07), 306 (Δ3.98), 290 (Δ3.70), 246 (Δ−5.26), 217 (Δ−6.16), 201 (Δ−6.01).
tribromide in CH2Cl2 (1 M, 4.8 cm3). After leaving the reaction mixture stirred, at room temperature overnight, it was cooled to −60 °C and water (5 cm3) was added and the mixture allowed to warm up to room temperature. The CH2Cl2 layer was separated and the remaining aqueous solution was extracted with EtOAc (2 × 10 cm3). The combined organics extract was dried (Na2SO4) and concentrated, under reduced pressure, to yield the crude catechol 29. Crystallisation from MeOH furnished catechol 29 as an off-white coloured solid (0.38 g, 85%); m.p. 211 °C (from MeOH); Rf 0.1 (50% EtOAc/hexane); (Found: M+ 309.1166. C17H17NO4 requires M+, 309.1158); δH (500 MHz, CDCl3) 1.62 (3 H, s, Me), 1.75 (3 H, s, Me), 4.43 (3 H, s, OMe), 5.01 (2 H, d, J=J1,J2=4.5, 3'-H), 7.50 (1 H, d, J=J3,J4=2.8, 3-H), 7.15 (1 H, d, J=J5,J6=9.2, 6-H), 7.57 (1 H, d, J=J3,J4=2.8, 2-H), 7.95 (1 H, d, J=J5,J6=9.2, 5-H); m/z (EI): 299 (M+, 20%), 231 (100).

iso-Haplopine-3,3′-dimethallylether 38. 4-Methoxy-8-[3-methyl-2-butenyl]oxufuro[2,3-b]quinolin-7-ol 37 (20 mg, 0.067 mmol) was treated with an excess of diazomethane as described earlier. The crude methylated product was purified by PLC to give compound 38 as a white solid (19 mg, 90%); m.p. 124–125 °C (from CH2Cl2); (lit.13 120–121.5 °C); Rf 0.4 (40% EtOAc/hexane); (Found: M+, 313.1312. C16H14NO4 requires 313.1314); δH (500 MHz, CDCl3) 1.66 (3 H, s, Me), 1.74 (3 H, s, Me), 4.00 (3 H, s, OMe), 4.43 (3 H, s, OMe), 4.84 (2 H, d, J=J1,J2=7.2, 1'-H), 5.73 (1 H, t, J=J2,J3=7.2, H-2'), 7.04 (1 H, d, J=J3,J4=2.8, 3-H), 7.22, d, J=J6,J7=4.6, 6-H), 7.58 (1 H, d, J=J3,J4=2.8, 2-H), 8.00 (1 H, d, J=J5,J6=9.4, 5-H); m/z (EI): 313 (M+, 20%).

Pteleine 40. cis-Dihydrodiol 16 (7 mg, 0.004 mmol) was dissolved in CDCl3 in an NMR tube and a drop of TFA added to the solution. The tube was kept at 50 °C until the aromatization of diol 16 was complete (1H-NMR analysis). After removal of solvent the aromatized product was treated with excess of diazomethane and the crude product obtained was purified by PLC (40% EtOAc/hexane) to afford pure pteleine 40 as a white solid (3 mg, 77%); m.p. 132–134 °C (lit.13 134.5 °C); Rf 0.4 (40% EtOAc/hexane); δH (500 MHz, CDCl3) 3.95 (3 H, s, Me), 4.60 (3 H, s, Me), 7.07 (1 H, d, J=J3,J4=2.8, 3-H), 7.36 (1 H, d, J=J7,J8=9.15, J=J5,J6=2.95, 7-H), 7.53 (1 H, d, J=J6,J7=2.95, 8-H), 7.63 (1 H, d, J=J3,J4=2.8, 2-H), 7.92 (1 H, d, J=J5,J6=9.15, 8-H).

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References


